# Observation of the surface 4f state of CePd<sub>7</sub> by means of the resonant-inverse-photoemission study at the Ce 4d absorption edge

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The resonant inverse photoemission study (RIPES) of CePd<sub>7</sub>, has been carried out at the Ce  $4d \rightarrow 4f$  absorption edge. The strong resonant enhancement of the 4f cross section enables us to distinguish two 4f components in the empty electronic state near the Fermi level. The incidence-angle dependence of the RIPES indicates a clear difference between ground-state configurations at the bulk and surface. It is found that the former shows a strongly hybridized 4f state, while the latter shows a localized 4f character. The angle dependence of the RIPES of  $\alpha$ -Ce metal has been also carried out and similar results as those of CePd<sub>7</sub> were obtained. The RIPES at the Ce  $4d \rightarrow 4f$  edge is found to be a powerful method to investigate the surface 4f state. [S0163-1829(97)07604-2]

### I. INTRODUCTION

It is known that Ce compounds show interesting electronic properties that are mainly caused by the partially localized character of the 4f electron. The high-energy spectroscopy has played an important role in the investigation of the electronic structure of these strongly correlated systems. Furthermore, an analysis of these measurements has achieved a great success within the framework of the degenerate impurity Anderson model (IAM).  $^1$ 

Some Ce compounds form the so-called valence fluctuation (VF) system, where the 4f electron has an itinerant character because of the strong hybridization with the conduction band. For typical VF systems, lower values of the Sommerfeld coefficient  $\gamma$  and the Pauli-like susceptibility  $\chi_0$  were reported. The 4f occupancies in the ground state of typical VF systems were shown to be about 0.8 by 3d x-ray photoemmision spectroscopy (XPS), x-ray absorption spectroscopy (XAS), photoemission spectroscopy (PES), and bremsstrahlung isochromat spectroscopy (BIS) measurements. Also the valence extracted from the XAS measurement at Ce 2p ( $L_{2,3}$ ) edge was usually found below 3.3.4 That is to say, the high-energy spectroscopy indicates that the ground state is mainly composed of  $4f^1$  configuration even in VF systems.

On one hand, CePd<sub>7</sub>, has the lowest values of  $\gamma$  (9.8 mJ K<sup>-2</sup> mol<sup>-1</sup>) (Ref. 4) and  $\chi_0$  (0.125×10<sup>-3</sup> emu mol<sup>-1</sup>) [Ref. 5(a)] as compared to typical VF systems. Furthermore, the smaller ratio  $\chi_0/\gamma$  (0.012 K<sup>2</sup> J<sup>-1</sup>) [Ref. 5(b)] is comparable with the case of nearly-free-electron systems. Moreover the valence number was determined to be 3.5 by XAS measurement at the  $L_3$  edge.<sup>5</sup> These facts suggest that CePd<sub>7</sub> is considered to belong to the strong hybridization limit: a very high Kondo temperature  $T_K$  which is a characteristic constant of the system, is expected. On the other hand, the com-

plicated structure of Ce- 3d XPS spectra suggests the strongly correlated nature of the electronic structure of CePd<sub>7</sub>. <sup>6</sup>

The crystal structure of CePd<sub>7</sub> has been confirmed to be an  $\alpha$ -Pd fcc structure with a lattice parameter of 4.002 Å. The Ce ion is surrounded by 12 nearest-neighbor Pd atoms. The direct f-f overlapping is negligibly small owing to a large Ce-Ce spacing ( $\sim$ 5.67 Å), and so, the 4f electron of CePd<sub>7</sub> might migrate mainly through the hybridization between Ce 4f and Pd valence states. Therefore, we can treat an 4f electron on one Ce site as almost isolated, and so IAM is adequate to describe approximately the electronic structure of CePd<sub>7</sub>.

The 4f-electronic structure at the outermost surface layers of the intermetallic Ce compounds is often different from that of bulk due to different hybridization strengths. Generally, the surface 4f state of the Ce compound is  $\gamma$  like. Recent analysis of the 2p XAS and 3d XPS spectra by Iwamoto et al.<sup>8</sup> reported that the occupancy of Ce 4f state at the surface and bulk of CePd $_7$  is 0.94 and 0.57, respectively. In marked contrast to the bulk 4f electron, the surface's 4f electron is well localized. This manifest difference of 4f-electronic structures may be reflected in the inverse photoemission spectroscopy (IPES) spectrum.

The IPES is known to be a useful measurement to investigate the unoccupied states. However, its cross section is much smaller than in the corresponding photoemission process. Moreover, the small magnitude of the 4f cross section prevents us from distinguishing the 4f signal from the non-f (for example, Ce 5d) signals and the large background in the IPES spectrum. Recently, several resonant IPES (RIPES) measurements of Ce compounds near the Ce 3d ( $M_5$ ) absorption edge around 883 eV have been performed. Only 10,111 They have demonstrated the great ability of the RIPES measurement to enhance the magnitude of only the 4f signal due to

the resonance effect. Therefore, the RIPES is known to be a powerful tool to investigate 4f-electron systems.

In this paper, the RIPES measurement has been performed near the Ce 4d ( $N_{4,5}$ ) absorption edge of CePd<sub>7</sub>. Two sharp peaks are resonantly enhanced near the Fermi level. These peaks show different excitation energy dependencies. Furthermore, the angle dependence of RIPES can resolve these components into bulk and surface states, because of its surface sensitivity.

### II. EXPERIMENT

The sample preparation and characterization have been presented elsewhere.  $^{5,6}$  Measurements were performed in an ultrahigh vacuum chamber where the base pressure is about  $7\times10^{-11}$  Torr. The CePd<sub>7</sub> sample was cleaned at liquid-N<sub>2</sub> temperature by scraping with a diamond file in a vacuum every 30 min. The Ce metal was evaporated to the molybdenum plate at 30 K. The surface cleanliness was checked by monitoring that there is no O 1s x-ray emission signal.

A filament-cathode-type electron gun was used for the excitation source. The kinetic energy  $(E_k)$  of the electron was calibrated by the electron energy analyzer. The IPES was measured by the soft x-ray emission spectrometer of the Institute for Solid State Physics, 12 which covers the wide photon energy range from 15 to 1200 eV. The spectrometer uses the Rowland circle geometry that consists of gratings with groove density of 300, 1200, 2400 lines/mm and the Cs-coated multichannel detector. We used a 300-lines/mm grating in this measurement. The Fermi-level position was determined by referring to the Fermi edge in the IPES spectra of Au which was evaporated on the same sample holder. The resolution of this system is 1.0 eV at  $E_k$ =124 eV.

## III. RESULTS AND DISCUSSION

Figure 1 shows the excitation energy dependence of the IPES spectra of CePd<sub>7</sub> for the incident electron energy in the Ce  $4d \rightarrow 4f$  region. The abscissa is the energy above the Fermi level that was calibrated by the Fermi edge of gold. The broad bands indicated by the vertical bars are the normal Ce 4d fluorescence that is caused by the Ce 4d core hole created by the electron excitation. The vertical bars located at higher energy show the fluorescence bands caused by the Ce  $5p \rightarrow 4d$  transition. It is found that two separate resonances near the Fermi level show conspicuous enhancement around  $E_k$ =124 eV. Obviously a peak at the lower-energy side, which we name the  $f^1$  state, mainly originates from the  $4f^0$ configuration in the ground state and the other, named the  $f^2$ state, originates from the  $4f^1$  configuration. The energy separation between the  $f^1$  peak and the Fermi level is about 0.95 eV, which reflects extremely high Kondo temperature (about  $1\times10^4$  K). The extremely high  $T_K$  of CePd<sub>7</sub> reflects the very strong hybridization between the 4f state and the conduction band. This result is consistent with the strong hybridization picture of the CePd<sub>7</sub> system, which is expected by the small values of the  $\gamma$  and  $\chi_0$ .

The normal IPES process is represented for the transition to Ce 4f states as follows:

$$|4d^{10}4f^{n}\rangle + e^{-} \rightarrow |4d^{10}4f^{n+1}\rangle + h\nu.$$
 (1)

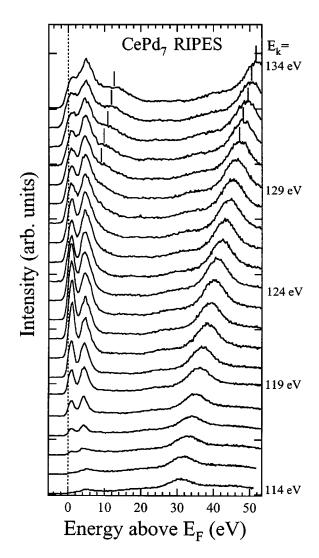


FIG. 1. The IPES spectra of  $CePd_7$ , near the  $Ce\ 4d \rightarrow 4f$  edge measured at liquid- $N_2$  temperature. The vertical dotted line shows the Fermi level. The abscissa is the energy above Fermi level  $(E_F)$ . The numbers written on the side of the right axis represent the kinetic energy  $(E_k)$  of the electron.

Here, n is the configuration number of the 4f electrons in the ground state. For the incident energy range of the present experiment, the contribution from non-f conduction bands also coexists with the 4f contribution in the normal IPES. In the RIPES experiment, on the other hand, the resonant processes are expressed by the following processes:

$$|4d^{10}4f^{n}\rangle + e^{-} \rightarrow |4d^{9}4f^{n+2}\rangle \rightarrow |4d^{10}4f^{n+1}\rangle + h\nu.$$
 (2)

Since the initial and final states are the same in these two processes (1) and (2), they interfere with each other. Therefore, the 4f cross section increases when the excitation energy is tuned to Ce  $4d \rightarrow 4f$  absorption edge and we can extract the Ce 4f contribution.

Figure 2 shows the comparison between the spectrum measured at  $E_k$ =114 eV and that at 126 eV. Their spectral intensities are normalized at the  $f^2$  peak intensities. The electron energy  $E_k$ =114 eV is just below the giant resonance of the Ce  $4d \rightarrow 4f$  absorption. The line shapes of the  $f^2$  peaks apparently are different from each other. At 114 eV, the  $f^2$  peak has a wider and asymmetric line shape as compared

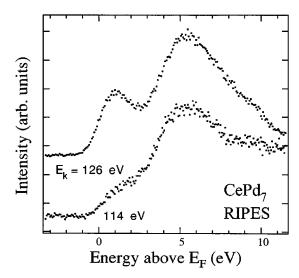


FIG. 2. Comparison between the IPES spectra near  $E_F$  measured at  $E_k$ =114 and 126 eV.

with that of 126 eV. As a possibility of the change of the line shape, it may reflect the band structure except the 4f component in CePd<sub>7</sub>, because the cross section of the 4f component is less below the resonance energy. In fact, the bandcalculation curve<sup>6</sup> well reproduces the line shape, if the energy position of the calculation is shifted to the higher energy by 1.4 eV. As another possibility, one can consider the change of line shape may reflect the selection rule on the resonance of the  $f^2$  state, since the Ce  $4d \rightarrow 4f$  absorption has several fine structures in the energy region below the giant resonance so that the  $f^2$  state causes the weak enhancement just below the giant resonance. In fact, the Ce  $5p\rightarrow4d$ fluorescence is observed at  $E_k=114$  eV, since the Ce  $4d \rightarrow 4f$  excitation has already occurred at this energy. Tanaka and Jo<sup>13</sup> have pointed out that the weakly enhanced  $f^2$  state just before the giant resonance has a wider line shape with several multiplet structures than that of the most enhanced one.

Figure 3 shows the excitation energy dependencies of the intensity of the  $f^1$  and  $f^2$  peaks. These curves are obtained by plotting the intensities of the  $f^1$  and  $f^2$  peaks against the kinetic energy of the electron, which corresponds to the constant final-state spectra. Each line shows the different energy dependence. The peak energy of the  $f^1$  curve is located at 124 eV and that of the  $f^2$  curve at 124–129 eV, and both have broad and asymmetric line shape. The line shape in Fig. 3 is completely different from the constant final-state spectra of the RIPES at the Ce 3d edges that were already measured for several Ce compounds. 10,11,13 This fact indicates much larger electronic interactions between the  $4f^n$  and 4d-core hole than those between the  $4f^n$  and 3d-core hole. Here one should remember that the 4d absorption spectrum has much wider and asymmetric line shape than that of the 3d absorption spectra in metallic Ce.4 In Ref. 13, it is pointed out that the line shapes of these curves in Fig. 3 are strongly influenced by the magnitude of the 4f-ligand hopping integral in the intermediate state at the resonance, taking account of the multiplet structures and the selection rule on resonance. Further discussion and full analysis of the energy dependence of RIPES spectra will be made in a separate paper.

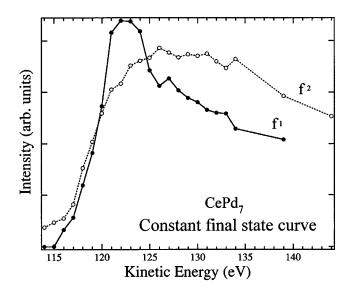


FIG. 3. The intensities of  $f^1$  and  $f^2$  peaks plotted against the kinetic energy of the electron.

The intensity ratio of the  $f^1/f^2$  peak in the Ce 4d RIPES spectrum of CePd<sub>7</sub> is much smaller than that of the Ce 3d RIPES result of CeRh<sub>3</sub>, one of the most hybridized systems, as reported by Grioni *et al.*<sup>11</sup> Here, one can find that the intensity ratio of the  $f^1/f^2$  peak on the resonance is unexpectedly small, supposing both states are identified with the bulk contribution.

It is said that the surface valence of rare-earth compounds is often lower than that of the bulk. In fact, it has been pointed out8 that the surface state shows a weak hybridization with a  $\gamma$ -like character in the analysis of 2p-XAS and 3d-XPS spectra of CePd<sub>7</sub> while the  $\alpha$ -like bulk state shows a strong hybridization. The mean free path of the electron, the kinetic energy that is around 100 eV, is about 5 Å, 14 so that the surface contribution strongly appears in the 4d RIPES spectrum. The angle dependence of the IPES spectra is useful to sort out both components. Higher sensitivity to surface is obtained by grazing incidence. Figure 4(a) shows the incidence-angle dependence of the RIPES measured at  $E_k = 124 \text{ eV}$  from  $0^{\circ}$  to  $80^{\circ}$ . The angle dependence was also measured at 118 eV, which is very similar to that in Fig. 4(a). It is clear that the intensity ratio of the  $f^1$  and  $f^2$  peaks increases by decreasing angles with decreasing sensitivity to the surface.

Iwamoto *et al.* calculated the BIS spectra of CePd<sub>7</sub>, where both the bulk and surface contributions are calculated independently and superposed with various weights. Figure 4(b) shows their results, where the relative intensity of the surface contribution is changed as a parameter. Though their calculation was limited to the off-resonance case, they took into account only the 4f contribution, so that their result should be compared with the experimental data of RIPES that reflects the 4f contribution. The  $f^1$  state of the calculated BIS spectrum at 0% surface contribution is in good agreement with our experimental spectrum. The angle dependence in Fig. 4(a) agrees well with that of the calculation of BIS spectra by Iwamoto *et al.* The linewidth of the  $f^1$  state is much wider in the experiment than in the calculation. A slight shoulder seems to be found at the Fermi edge for the

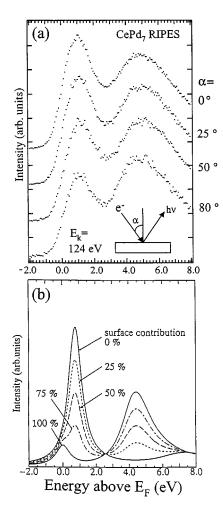


FIG. 4. (a) The angle dependence of the on-resonant RIPES spectra of  $CePd_7$  measured at  $E_k$ =124 eV. (b) The calculation (Ref. 8) of the BIS spectra of  $CePd_7$  where the surface contribution changes from 0% to 100%.

spectrum at the  $80^{\circ}$  incidence angle. The predicted doublet structures of the  $f^1$  state at 75% surface contribution are not clearly resolved in the experiment because of its wide linewidth. The surface contributions in the experimental spectra from  $0^{\circ}$  to  $80^{\circ}$  correspond to be in the range from 50% to 75% in the calculations.

The surface contribution in the RIPES spectrum has been calculated by the following simple formula:<sup>15</sup>

$$I_s/(I_s + I_h) = 1 - \exp(-a/l \cos \alpha),$$
 (3)

where  $I_s$  and  $I_b$  are the surface and the bulk state intensities, respectively, l is the mean free path of the electron,  $\alpha$  is the incidence angle, and a is the thickness of surface layer. Here, l is assumed to be 5 Å for the electron with the kinetic energy of about 100 eV and a is assumed to be 2 Å, which corresponds to the surface monolayer of CePd<sub>7</sub>. Figure 5 shows the surface contribution of Ce 4d and Ce 3d edges in the spectra against the incidence angle. The incidence angle from  $0^{\circ}$  to  $80^{\circ}$  correspond to the surface contribution from 40% to 90% in Fig. 5. It is in very good agreement with a rough estimation of the surface contribution from 50% to

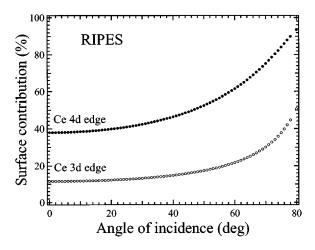


FIG. 5. The angle dependence of the surface contributions in the Ce 4d and 3d RIPES spectra.

75%, which is obtained by the comparison between the calculated and the experimental line shapes between the  $f^1$  and  $f^2$  states, as shown in Fig. 4.

It is adequate to consider that the  $f^1$  and  $f^2$  peaks are composed of the bulk and the surface contribution, respectively. Therefore, it is concluded that the high  $T_K$  extracted from the  $f^1$  peak position represents the extremely strong hybridization of the bulk 4f state and the  $f^2$  peak comes from the surface 4f state. The different dependence on the excitation energy of the  $f^1$  and  $f^2$  peaks indicates the very contrastive electronic structure; that is, the bulk is  $\alpha$ -Ce-like and the surface is  $\gamma$ -Ce-like. For more detailed discussions on the RIPES spectra (e.g., the incident energy dependence), the calculation of the RIPES will be made in the near future taking into account processes (1) and (2).

As another example of the angle dependence of the RIPES, Fig. 6 shows the angle dependence of the on-resonance spectra of  $\alpha$ -Ce metal. The intensity of the  $f^1$  state

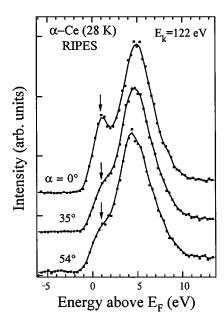


FIG. 6. The angle dependence of the on-resonant RIPES spectra of  $\alpha$ -Ce metal.

indicated by arrows increases from grazing to normal incidence angle, while the  $f^2$  state remains constant for each angle. Since there is a similar angle dependence of the spectra as in the case of CePd<sub>7</sub>, it is found that the  $\gamma$ -like surface state also exists in the  $\alpha$ -Ce metal. It may not be particular that the more localized state exists at the surface state of the Ce compounds. The RIPES is a powerful experimental method to study such 4f states.

Figure 5 shows the comparison of the angle-dependent surface contribution between the 4d and 3d RIPES. The Ce 4d RIPES is more surface sensitive than 3d RIPES, because of its short electron mean free path. Its angle dependence is also larger, so that it may be more powerful experimental method to measure the surface contribution in the 4d RIPES than in the 3d RIPES.

### IV. CONCLUSION

In conclusion, the RIPES measurement near the Ce 4d absorption edge reveals the remarkable contrastive electronic structure at the bulk and surface state of CePd<sub>7</sub>. The RIPES

at the Ce  $4d \rightarrow 4f$  edge is found to be a powerful method to investigate the surface 4f state. The bulk is  $\alpha$ -like and the surface is  $\gamma$ -like. The difference of the ground-state configuration at the bulk and surface reflects the separate resonances. The extremely high Kondo temperature  $(kT_K \sim 0.95 \, \text{eV})$  of the bulk expresses the very strong hybridization between the 4f and ligand states. Our results are consistent with the previous calculation, though more detailed calculations will be needed. Due to the large Ce-Ce spacing, the calculation within the framework of the IAM seems to give a good agreement. The resonant enhancement of the 4f cross section makes it possible to distinguish between the specific contributions in the spectra near Fermi level. It is pointed out that the RIPES measurement around Ce 4d edge has a high sensitivity to the surface.

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<sup>&</sup>lt;sup>1</sup>O. Gunnarsson and K. Schönhammer, Z. Phys. B **30**, 297 (1978); Phys. Rev. B **28**, 4315 (1983).

<sup>&</sup>lt;sup>2</sup>J. G. Sereni, in *Handbook on the Physics and Chemistry of Rare Earths*, edited by K. A. Gschneider, Jr. and L. Eyring (North-Holland, Amsterdam, 1991), Vol. XV, Chap. 98.

<sup>&</sup>lt;sup>3</sup> J. W. Allen, S. J. Oh, O. Gunnarsson, K. Schönhammer, M. B. Maple, M. S. Torikachvili, and I. Lindau, Adv. Phys. **35**, 275 (1986); J. C. Fuggle, F. U. Hillebrecht, Z. Zolnierek, R. Lässer, Ch. Freiburg, O. Gunnarsson, and K. Schönhammer, Phys. Rev. B **27**, 7330 (1983).

<sup>&</sup>lt;sup>4</sup>J. Röhler, in *Handbook on the Physics and Chemistry of Rare Earths*, edited by K. A. Gschneider, Jr. and L. Eyring (North-Holland, Amsterdam, 1987), Vol. X, Chap. 71; J. G. Sereni, O. Trovarelli, J. Schaf, G. Schmerber, and J. P. Kappler, Mod. Phys. Lett. **5**, 1249 (1991).

<sup>&</sup>lt;sup>5</sup> (a) H. Takeya, K. A. Geshneider, B. J. Beaudry, T. Ellis, and V. K. Pecharsky, J. Alloys Compd. **209**, 35 (1994); (b) J. G. Sereni, O. Trovarelli, A. Herr, J. Ph. Schillé, E. Beaurepaire, and J. P. Kappler, J. Phys. Condens. Matter **5**, 2927 (1993).

<sup>&</sup>lt;sup>6</sup>E. Beaurepaire, J. P. Kappler, S. Lewonczuk, J. Ringeissen, M. A.

Khan, J. C. Parlebas, Y. Iwamoto, and A. Kotani, J. Phys. Condens. Matter 5, 5841 (1993).

<sup>&</sup>lt;sup>7</sup>D. A. Smith, I. P. Jones, and I. R. Harris, J. Mater. Sci. Lett. 1, 463 (1982).

<sup>&</sup>lt;sup>8</sup> Y. Iwamoto, M. Nakazawa, A. Kotani, and J. C. Parlebas, J. Phys. Condens. Matter 7, 1149 (1995).

<sup>&</sup>lt;sup>9</sup>F. U. Hillebrecht and M. Campagna, in *Handbook on the Physics* and *Chemistry of Rare Earths* (Ref. 4), Chap. 70.

<sup>&</sup>lt;sup>10</sup>P. Weibel, M. Grioni, D. Malterre, B. Dardel, and Y. Baer, Phys. Rev. Lett. **72**, 1252 (1994).

<sup>&</sup>lt;sup>11</sup>M. Grioni, P. Weibel, D. Malterre, F. Jeanneret, Y. Baer, and G. Olcese, Physica B 206 & 207, 71 (1995).

<sup>&</sup>lt;sup>12</sup>S. Shin, A. Agui, M. Fujisawa, Y. Tezuka, T. Ishii, and N. Hirai, Rev. Sci. Instrum. **66**, 1584 (1995).

<sup>&</sup>lt;sup>13</sup> A. Tanaka and T. Jo, Physica B **206** & **207**, 74 (1995); J. Phys. Soc. Jpn. **65**, 615 (1996).

<sup>&</sup>lt;sup>14</sup>F. Gerken, A. S. Flodstrm, J. Barth, L. I. Johansson, and C. Kunz, Phys. Scr. **32**, 43 (1985).

<sup>&</sup>lt;sup>15</sup>M. H. Hecht, A. J. Viescas, I. Lindau, J. W. Allen, and L. I. Johansson, J. Electron Spectrosc. Relat. Phenom. 34, 343 (1984).